Understanding Lithium-ion Batteries – Cryo and *in situ* TEM for Materials Science

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Battery Applications







Battery Applications



Safety Issues & Degradation

Intended failure of LiPo battery



Real battery

Charging Deposition of Li at anode Li-dendrite formation Internal short circuit



NewsD2 & Oak Ridge National Laboratory/Youtube.com

Extreme Reactivity of Lithium

- Lightest solid He element/metal 1.008 Alkalimetalle -0,00(1) 6,94 2.20 10 20,180 bergangsmetall 0.5 g/cm³ Lanthanoide -259/-253 Be Bor Ne п - One of the most 12 24, 13 18 39,948 Wasserstoff Na D ς Cl Mg Al Ar reactive elements 19 39,098 20 40,078 72,63 33 74,922 35 79,9 36 83,798 Mn 7.4.4 Fe 4.3.2 Kn 7.4.4 53 / 2732 3, 2, 0 Κ Ga Ge As Se Br Kr Ca 37 85,468 Li in air Li in water Li + water \rightarrow LiOH Li + air \rightarrow Li₂O, Li₃N

Science247/TeacherTube.com

Thoisoi2/Youtube.com

Targeted Materials Development

Preventing battery failure Local properties, morphology, structure and chemistry of Li dendrites, cell parts, interfaces

High-capacity batteries with <u>pure-Li anode</u>



Characterization on very small length scales down to the <u>nanometer</u> and <u>atomic scale</u>





Size Scales — How small is a Nanometer?



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How small is an Atom?





1 The Tip of a Pencil 10,000,000,000









0,1 nm Atom

Hitachi/Youtube.com

How to Probe the nm-Level — LM vs. TEM

Resolution (Ernst Abbe)

λ=[400 nm, 800 nm]



0.0

Insind

leica-microsystems.com

Microscope

Electron

Transmission



Fast electrons (0.8 c)

 $\lambda_{\text{deBroglie}}$ (pm = 10⁻¹² m) 2 pm @ 300 keV

FEI Titan Themis

Light Microscope — TEM



Probing Materials at the nm- and Atomic Level



Probing Battery Materials at the nm-Level

Characterization of Li dendrites, cell parts, interfaces
 Systematic study of reaction of Li with O₂, N₂, H₂O

Regular TEM sample handling in air & TEM analyses in high vacuum

→ How to investigate reactive Li and battery parts?
→ How to study Li reactions (corrosion, passivation)



Li electron-beam sensitive!





Probing Battery Materials at the nm-Level



Li et al, Science 358 (2017) 506

 Battery disassembly
 + TEM sample preparation in Ar environment

> TEM characterization at -196 °C (LN₂) **Reduced Li <u>reactivity</u> & Li <u>mobility</u> (+ enhanced vacuum)**

3) Transfer w/





4) TEM characterization under cryo-conditions



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Characteristics of Li-metal dendrites @ -196 °C

Single-crystalline dendrites



Atomic-resolution TEM



Cryo-TEM allows for detailed characterization of sensitive battery materials but we are still in vacuum



(K-edge)

Mosaic-type model

Solid

electrolyte interface (SEI)

5 nm

Li et al, Science 358 (2017) 506

Complex multiphase SEI

*i*Li {110}

Reactions of Metallic Li by in situ TEM



Understanding Li-ion Batteries

Passivation of Li in dry N_2

6Li + $N_2 \rightarrow 2Li_3N$ (at Li/Li₃N interface)



Passivation of Li in dry N_2

6Li + $N_2 \rightarrow 2Li_3N$ (at Li/Li₃N interface)





Passivation of Li in dry N₂

Video not available

Corrosion of Li in humid environment (air)





 $6Li + N_2 \rightarrow 2Li_3N$ $2Li + 2H_2O \rightarrow 2LiOH + H_2$ $Li_3N + 3H_2O \rightarrow 3LiOH + NH_3$ Gaseous reaction products \rightarrow Pores

Video not available

Reactions of Metallic Li \rightarrow Battery Development



→ Novel Li-ion batteries with <u>Li-metal anode</u> + <u>passivation</u> <u>interlayer</u> with high capacity and long-term cycling stability

Lessons learned

TEM contributes to development of <u>safe</u>, <u>durable</u>, <u>light</u> batteries with <u>high capacity</u> & <u>fast charging</u>

Establishing lowdose microscopy for materials science



in situ & environmental TEM Characterzation of reactions of sensitive battery materials



Understanding Li-ion Batteries

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Li, Li, Pei, Yan, Sun, Wu, Joubert, Chin, Koh, Yu, Perrino, <u>Butz</u>, Chu, Cui Atomic structure of sensitive battery materials and interfaces revealed by cryo-electron microscopy Science 358 (2017) 506

Li, Li, Sun, <u>Butz</u>, Yan, Koh, Zhao, Pei, Cui **Revealing Nanoscale Passivation and Corrosion Mechanisms of Reactive Battery Materials in Gas Environments** Nano Letters 17 (2017) 5171

